

Stereospecific Cross-Coupling of Secondary Alkyl β -Trifluoroboratoamides

Deidre L. Sandrock,^{†,§} Ludivine Jean-Gérard,^{†,§} Cheng-yi Chen,[‡] Spencer D. Dreher,[‡] and Gary A. Molander^{*,†}

Roy and Diana Vagelos Laboratories, Penn/Merck Laboratory for High-Throughput Experimentation, Department of Chemistry, University of Pennsylvania, 231 South 34th Street, Philadelphia, Pennsylvania 19104-6323, United States, and Department of Process Chemistry, Merck Research Laboratories, P.O. Box 2000, Rahway, New Jersey 07065, United States

Received October 4, 2010; E-mail: gmolandr@sas.upenn.edu

Abstract: The stereospecific cross-coupling of enantioenriched *nonbenzylic* secondary alkyl boron compounds has been achieved. The high selectivity toward product formation over an undesired β -H elimination pathway is achieved via an intramolecular coordination of an ancillary carbonyl to the metal center in the diorganopalladium intermediate.

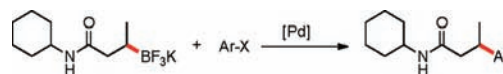
The Suzuki–Miyaura cross-coupling reaction has emerged as one of the most versatile transformations available for the generation of C–C bonds.¹ Although there are many strategies that exist for the cross-coupling of sp^2 -hybridized organometallics,² the protocols for the cross-coupling of secondary and potentially enantioenriched sp^3 -hybridized organometallics have limited precedent.³ In a previous communication, we described a development toward this goal by identifying catalytic reaction conditions for the cross-coupling of cyclic, symmetrical secondary alkyltrifluoroborates with aryl electrophiles.⁴ However, when applied to symmetrical acyclic substrates, it became evident that the use of our optimized conditions still led to a β -H elimination/isomerization pathway that resulted in mixtures of the desired cross-coupled products as well as the undesired isomerized primary alkylated products.

More recently, other attempts at secondary cross-coupling using various organoboron derivatives have appeared.⁵ Crudden and co-workers developed a protocol for the preparation of enantioenriched styrene-derived secondary boronate esters and demonstrated their cross-coupling with a variety of aryl iodides.⁵ Although the reactions proceed in good yields with *retention* of configuration,⁶ this method was limited to the cross-coupling of benzylic boron derivatives.

During the course of this investigation, Suginome reported the cross-coupling of α -(acylamino)benzylboronates with aryl bromides and chlorides, which somewhat surprisingly occurred with overall *inversion* of configuration.⁷

Herein we report our most recent efforts toward the ultimate goal of cross-coupling *nonbenzylic*, enantioenriched secondary alkyl organoboron reagents with stereochemical fidelity during the cross-coupling event. Subsequent to our studies on secondary alkyltrifluoroborates, efforts were conducted to extend the study of β -trifluoroborato carbonyls⁹ to the preparation and cross-coupling of acyclic secondary β -trifluoroboratoamides. Using the borylation strategy outlined by Yun and co-workers, a variety of these trifluoroborates were prepared.^{9a,10} With the desired substrates in hand, an initial screen of catalytic conditions led to the combination of 10 mol % of Pd(OAc)₂ and 20 mol % of XPhos, K₂CO₃ in a

Table 1. Cross-Coupling of β -Trifluoroboratoamides with Aryl Halides^a



entry	electrophile	X	product	% isolated yield
1		Cl Br		90 63
2		Cl Br		72 74 ^b
3		Cl		<i>o</i> = 88 <i>p</i> = 91
5		Cl		81
6		Cl Br		76 72 ^b
7		Cl Br		66 ^b 47 ^b
8		Cl		72
9		Br		92
10		Cl		83
11		Cl		80 ^b
12		Cl		92 ^b
13		Cl		71 ^b

^a General conditions: Pd(OAc)₂ (10 mol %), XPhos^{8a} (20 mol %), RBF₃K (1 equiv), K₂CO₃ (3 equiv), and 6.7:1 CPME/H₂O (0.25 M).

^b Reactions perform better with SPhos^{8b} (20 mol %) and Cs₂CO₃ (3 equiv).

cyclopentyl methyl ether (CPME)/H₂O solvent system giving the highest isolated yield of potassium *N*-cyclohexyl-3-(trifluoroborato)butanamide in the coupling reaction with 2-chloroanisole (Table 1, entry 1).

[†] University of Pennsylvania.

[‡] Merck Research Laboratories.

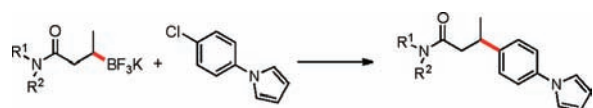
[§] These authors contributed equally.

Encouraged by this initial result, we proceeded with further screening, which revealed that the combination of 10 mol % of XPhos or SPhos with Cs₂CO₃ (3 equiv) also provided good to excellent yields of the cross-coupled products with both aryl chlorides and bromides.

Using 10 mol % of Pd(OAc)₂ and 20 mol % of XPhos, a variety of electrophilic partners (including those containing ketone, aldehyde, ester, nitrile, and nitro groups) cross-coupled with the model trifluoroborate in good yields. In a number of cases, the use of SPhos as the ligand actually provided higher yields of the cross-coupled product (Table 1, entries 6, 7, 11–13). In all of these examples, <2% of products resulting from β -H elimination or isomerization were isolated.

To investigate the scope of this reaction with respect to the nucleophilic partner, all three sets of suitable catalytic conditions were applied to a variety of amide substrates, in each case generating the cross-coupled products in good yields (Table 2, entries 1–7), again observing little or none of the undesired byproducts.

Table 2. Cross-Coupling of Various Trifluoroborates with Aryl Halides^a



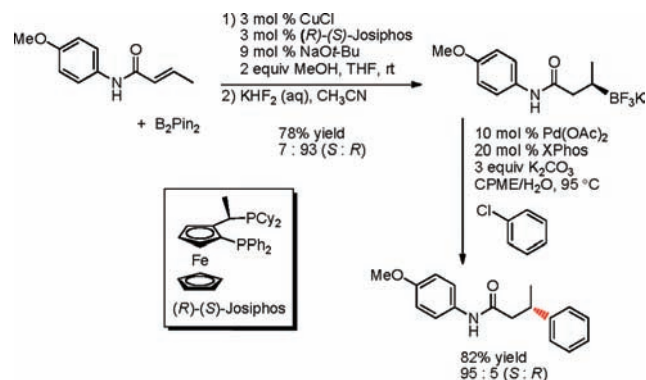
entry	RBF ₃ K	ligand/base	% isolated yield
1		XPhos, K ₂ CO ₃	84
		XPhos, Cs ₂ CO ₃	89
		SPhos, Cs ₂ CO ₃	78
2		XPhos, K ₂ CO ₃	92
		XPhos, Cs ₂ CO ₃	84
		SPhos, Cs ₂ CO ₃	93
3		XPhos, K ₂ CO ₃	87
		XPhos, Cs ₂ CO ₃	99
		SPhos, Cs ₂ CO ₃	98
4		XPhos, K ₂ CO ₃	72
		XPhos, Cs ₂ CO ₃	78
		SPhos, Cs ₂ CO ₃	64
5		XPhos, K ₂ CO ₃	89
		XPhos, Cs ₂ CO ₃	96
		SPhos, Cs ₂ CO ₃	89
6		XPhos, K ₂ CO ₃	79
		XPhos, Cs ₂ CO ₃	91
		SPhos, Cs ₂ CO ₃	86
7		XPhos, K ₂ CO ₃	94
		XPhos, Cs ₂ CO ₃	79
		SPhos, Cs ₂ CO ₃	89

^a General conditions: Pd(OAc)₂ (10 mol %), ligand (20 mol %), RBF₃K (1 equiv), base (3 equiv), and 6.7:1 CPME/H₂O (0.25 M).

With the ultimate goal of developing conditions to generate optically active materials through the use of an appropriate organoboron reagent, we prepared an enantioenriched β -trifluoroboratoamide via an asymmetric β -borylation reaction of the corresponding α,β -unsaturated amide using bis(pinacolato)diboron and (*R*)-(*S*)-Josiphos as the chiral ligand (Scheme 1).¹¹

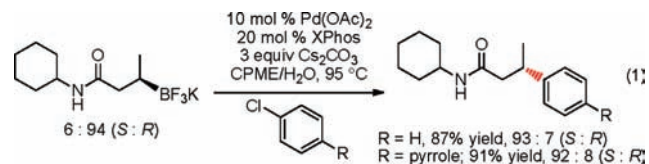
With the enantioenriched secondary organotrifluoroborate in hand, we subjected it to the optimized reaction conditions for the cross-coupling of this family of substrates. Using 10 mol % of Pd(OAc)₂, 20 mol % of XPhos, and 3 equiv of K₂CO₃ in a CPME/H₂O solvent system, the cross-coupled product was obtained in an enantiomeric ratio of 95:5 (*S*:*R*) in 82% yield for the cross-coupling

Scheme 1. Preparation and Cross-Coupling of Enantioenriched β -Trifluoroboratoamide



step. The absolute configurations of the major enantiomers of the borylated starting material and cross-coupled products were determined to be *R* and *S*, respectively, by comparison with the authentic *S* isomers prepared from derivatization of commercially available (*S*)-3-hydroxybutyric acid and (*S*)-3-phenylbutyric acid. This complete inversion in stereochemistry during transmetalation for secondary alkyl boron compounds (in substrates that have the potential for β -hydride elimination) represents an important extension to the previously described methods for the cross-coupling of secondary organometallics.

Subsequent cross-couplings with the enantioenriched cyclohexyl amide derivative with aryl chlorides also revealed the same inversion of configuration with no discernible stereochemical erosion detected (eq 1). Interestingly, neither the analogous β -trifluoroborato ketones nor -esters afford the desired coupled products.

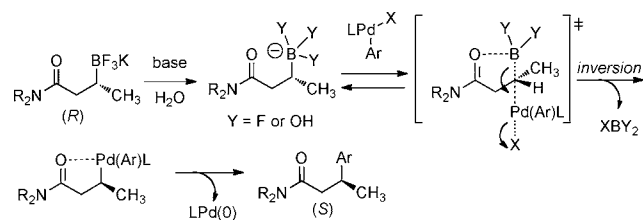


Although other factors could conceivably be involved, the unique reactivity of β -trifluoroboratoamides supports an hypothesis in which the ancillary carbonyl oxygen plays a role in coordinating with the intermediate diorganopalladium complex. Three beneficial features would derive from this interaction: (1) The coordination could facilitate the transmetalation process, as the conditions optimized for this transformation were not optimal for the cross-coupling of unfunctionalized secondary alkyltrifluoroborates described in our previous communication.⁴ (2) The complexation may also restrict the conformation of the diorganopalladium intermediate, inhibiting a syn-coplanar arrangement of the palladium and the acidic hydrogens alpha to the carbonyl required for β -hydride elimination.¹² (3) More importantly, the carbonyl interaction with the coordinatively unsaturated palladium could inhibit the metal from interacting agostically with the β -hydrogens, a feature required for β -H elimination (Scheme 2). These characteristics result in the formulation of a new paradigm for successful secondary alkyl cross-coupling with potentially wide implications.

As in the Sugimoto study, the inversion of configuration observed during the cross-coupling reaction with the β -trifluoroboratoamides is attributed to intramolecular coordination of the carbonyl group to the boron. Chiral benzylstannanes,¹³ silanes,¹⁴ and α -(acylamino)benzylboronic esters⁷ have been shown to undergo transmetalation with inversion of configuration, presumably through an S_E2 mechanism via an open transition state, a process

that is favored in polar solvents. More closely related to the current studies, examples of S_E2 -type reactions that proceed with inversion of configuration in borate substrates have been reported previously as well.¹⁵

Scheme 2. Proposed Mechanism for Complete Stereochemical Inversion



In conclusion, the concept of using pendant ligands to serve as hemilabile ligands¹⁶ to enhance transmetalation and inhibit the β -hydride elimination pathway in the cross-coupling of secondary organometallic species is highlighted. Additionally, the first cross-coupling of a nonbenzylic, enantioenriched secondary alkyl organometallic containing β -hydrogens that proceeds with complete inversion of configuration without any loss of enantioselectivity during the cross-coupling event has been reported.

Acknowledgment. G.A.M. thanks the NIH General Medical Sciences (GM035249) and the NSF GOALI program (CHE-0848460) for their generous support of this research. Dr. Rakesh Kohli (University of Pennsylvania) is acknowledged for obtaining HRMS data. BASF is acknowledged for their generous donation bis(pinacolato)diboron, and Solvias for the donation of (R)-(S)-JosiPhos. Petr Valenta, Jason Melvin, and Genette McGrew (University of Pennsylvania) are acknowledged for their assistance in HPLC and SFC analysis.

Supporting Information Available: Experimental details and spectral data of all compounds synthesized. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- (1) For reviews on potassium organotrifluoroborates, see: (a) Darses, S.; Genêt, J.-P. *Eur. J. Org. Chem.* **2003**, 4313. (b) Molander, G. A.; Figueroa, R. *Aldrichimica Acta* **2005**, 38, 49. (c) Molander, G. A.; Ellis, N. *Acc. Chem. Res.* **2007**, 40, 275. (d) Stefani, H. A.; Cella, R.; Adriano, S. *Tetrahedron* **2007**, 63, 3623. (e) Darses, S.; Genêt, J.-P. *Chem. Rev.* **2008**, 108, 288. (f) Molander, G. A.; Canturk, B. *Angew. Chem., Int. Ed.* **2009**, 48, 9240.
- (2) Kumada coupling: (a) Tamao, K.; Kiso, Y.; Sumitani, K.; Kumada, M. *J. Am. Chem. Soc.* **1972**, 94, 9268. (b) Nakamura, N.; Matsuo, K.; Ito, S.;

- (c) Nakamura, E. *J. Am. Chem. Soc.* **2004**, 126, 3686. (c) Nagano, T.; Hayashi, T. *Org. Lett.* **2004**, 6, 1297. (d) Bedford, R. B.; Bruce, D. W.; Frost, R. M.; Goodby, J. W.; Hird, M. *Chem. Commun.* **2004**, 2822. (e) Bedford, R. B.; Bruce, D. W.; Frost, R. M.; Hird, M. *Chem. Commun.* **2005**, 4161. (f) Bedford, R. B.; Betham, M.; Bruce, D. W.; Danopoulos, A. A.; Frost, R. M.; Hird, M. *J. Org. Chem.* **2006**, 71, 1104. (g) Ohmiya, H.; Yorimitsu, H.; Oshima, K. *J. Am. Chem. Soc.* **2006**, 128, 1886. (h) Bica, K.; Gaertner, P. *Org. Lett.* **2006**, 8, 733. (i) Bedford, R. B.; Betham, M.; Bruce, D. W.; Davis, S. A.; Frost, R. M.; Hird, M. *Chem. Commun.* **2006**, 1398. (j) Chowdhury, R. R.; Crane, A. K.; Fowler, C.; Kwong, P.; Kozak, C. M. *Chem. Commun.* **2008**, 94. Suzuki coupling: (k) Brenstrum, T.; Gerristma, D. A.; Adjabeng, G. M.; Frampton, C. S.; Britten, J.; Robertson, A. J.; McNulty, J.; Capretta, A. *J. Org. Chem.* **2004**, 69-7635. (l) Gonzalez-Bobes, F.; Fu, G. C. *J. Am. Chem. Soc.* **2006**, 128, 5360. Stille coupling: (m) Powell, D. A.; Maki, T.; Fu, G. C. *J. Am. Chem. Soc.* **2005**, 127, 510. Negishi coupling: (n) Nakamura, M.; Ito, S.; Matsuo, K.; Nakamura, E. *Synlett* **2005**, 1794. Hiyama coupling: (o) Strotman, N. A.; Sommer, S.; Fu, G. C. *Angew. Chem., Int. Ed.* **2007**, 46, 3556.
- (3) (a) Boudier, A.; Knochel, P. *Tetrahedron Lett.* **1999**, 40, 687. (b) Vyvyan, J. R.; Loitz, C.; Looper, R. E.; Mattingly, C. S.; Peterson, E. A.; Staben, S. T. *J. Org. Chem.* **2004**, 69, 2461. (c) Luo, X.; Zhang, H.; Duan, H.; Liu, Q.; Shu, L.; Zhang, T.; Lei, A. *Org. Lett.* **2007**, 9, 4571. (d) Miyaura, N.; Ishiyama, T.; Sasaki, H.; Ishikawa, M.; Satoh, M.; Suzuki, A. *J. Am. Chem. Soc.* **1989**, 111, 314. (e) Littke, A. F.; Dai, C.; Fu, G. C. *J. Am. Chem. Soc.* **2000**, 122, 4020. (f) Kataoka, N.; Shelby, Q.; Stambuli, J. P.; Hartwig, J. F. *J. Org. Chem.* **2002**, 67, 5553. (g) Hayashi, T. *Organomet. Chem.* **2002**, 653, 41. (h) Cahiez, G.; Marquais, S. *Tetrahedron Lett.* **1996**, 37, 1773. (i) Nakao, Y.; Takeda, M.; Matsumoto, T.; Hiyama, T. *Angew. Chem., Int. Ed.* **2010**, 49, 4447.
- (4) Dreher, S. D.; Dormer, P. G.; Sandrock, D. L.; Molander, G. A. *J. Am. Chem. Soc.* **2008**, 130, 9257.
- (5) (a) van den Hoogenband, A.; Lange, J. H. M.; Terpstra, J. W.; Koch, M.; Visser, G. M.; Visser, M.; Korstanje, T. J.; Jastrzebski, J. T. B. H. *Tetrahedron Lett.* **2009**, 49, 4122. (b) Ohmura, T.; Awano, T.; Suginome, M. *Chem. Lett.* **2009**, 38, 664. (c) Endo, K.; Ohkubo, T.; Hirokami, M.; Shibata, T. *J. Am. Chem. Soc.* **2010**, 132, 11033. (d) Imao, D.; Glasspoole, B. W.; Laberge, V. S.; Crudden, C. M. *J. Am. Chem. Soc.* **2009**, 131, 5024.
- (6) (a) Matos, K.; Soderquist, J. A. *J. Org. Chem.* **1998**, 63, 461. (b) Ridgway, B. H.; Woerpel, K. A. *J. Org. Chem.* **1998**, 63, 458.
- (7) Ohmura, T.; Awano, T.; Suginome, M. *J. Am. Chem. Soc.* **2010**, 132, 13191.
- (8) (a) XPhos (2-dicyclohexylphosphino-2',4',6'-diisopropyl-1,1'-biphenyl). (b) SPhos (2-dicyclohexylphosphino-2',6'-dimethoxybiphenyl).
- (9) (a) Molander, G. A.; Petrillo, D. E. *Org. Lett.* **2008**, 10, 1795. (b) Molander, G. A.; Jean-Gérard, L. *J. Org. Chem.* **2009**, 74, 1297. (c) Molander, G. A.; Jean-Gérard, L. *J. Org. Chem.* **2009**, 74, 5446.
- (10) (a) Mun, S.; Lee, J.-E.; Yun, J. *Org. Lett.* **2006**, 8, 4887. See also: (b) Bonet, A.; Gulyás, H.; Koshevoy, I. O.; Estevan, F.; Sanaú, Ubeda, M. A.; Fernández, E. *Chem.—Eur. J.* **2010**, 16, 6382. (c) Smith, S. M.; Takacs, J. M. *J. Am. Chem. Soc.* **2010**, 132, 1740.
- (11) Chea, H.; Sim, H.-S.; Yun, J. *Adv. Synth. Catal.* **2009**, 351, 855.
- (12) Spessard, G. O.; Miessler, G. L. *Organometallic Chemistry*; Prentice Hall: Upper Saddle River, NJ, 1997.
- (13) (a) Labadie, J. W.; Stille, J. K. *J. Am. Chem. Soc.* **1983**, 105, 6129. (b) Kells, K. W.; Chong, J. M. *J. Am. Chem. Soc.* **2004**, 126, 15666.
- (14) Hatanaka, Y.; Hiyama, T. *J. Am. Chem. Soc.* **1990**, 112, 7793.
- (15) (a) Bergbreiter, D. E.; Rainville, D. P. *J. Organomet. Chem.* **1976**, 121, 19. (b) Kabalka, G. W.; Gooch, E. E., III. *J. Org. Chem.* **1980**, 45, 3578.
- (16) (a) Giovannini, R.; Stüdemann, T.; Dussin, G.; Knochel, P. *Angew. Chem., Int. Ed.* **1998**, 37, 2387. (b) Johnson, J. B.; Rovis, T. *Angew. Chem., Int. Ed.* **2008**, 47, 840. (c) Luo, X.; Zhang, H.; Duan, H.; Liu, Q.; Zhu, L.; Zhang, T.; Lei, A. *Org. Lett.* **2007**, 9, 4571.

JA108949W